

EM/K-19600/A

Epoxy acrylates

11/21/03
The present invention relates to ~~novel~~ epoxy acrylates of higher molecular weight and to novel carboxyl group-containing epoxy acrylates of higher molecular weight, to the preparation thereof, to the use of said epoxy acrylates in photoresist formulations, and to the use of said formulations, in particular in the field of printed wiring boards, typically as solder resists or as primary resists (etch resists or galvanoresists), and of printing plates.

Epoxy acrylates are known in abundance and are also used, inter alia, in compositions used as photoresist formulations.

Formulations for solder resists that contain reaction products of epoxy novolak resins with acrylic acid and cyclic carboxylic anhydrides are disclosed, inter alia, in EP 0 273 729. They are developable in aqueous alkaline media and have good thermal resistance and photosensitivity. Their resistance to chemicals, however, is unsatisfactory.

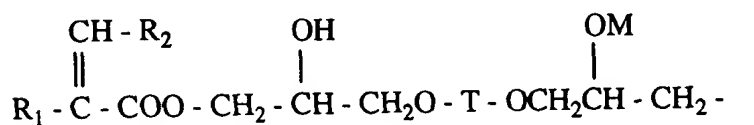
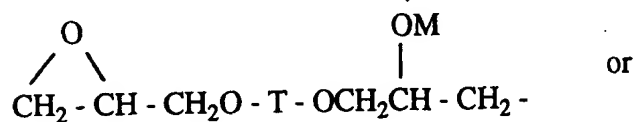
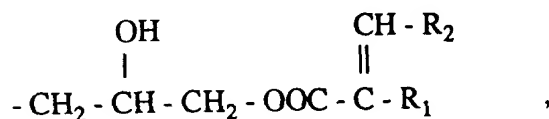
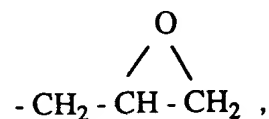
EP 0 418 011 discloses compositions for solder mask that are likewise based on reaction products of epoxy cresol novolaks with acrylic acid and cyclic dicarboxylic anhydrides, using 0.4 to 0.9 equivalent of acrylic acid per equivalent of epoxy group, such that the final product simultaneously contains acid and epoxy groups in the same molecule. A second thermal crosslinking reaction between these two functionalities is thereby made possible in the application of these resist compositions. The problem here is, however, aside from the preparation of the products (danger of gelation in the reaction with the anhydride), the storage stability, as the formulation containing such reaction products has a certain reactivity even at room temperature. All these cited epoxy acrylates are quite generally relatively low-molecular.

Photochemically or thermally cured epoxy acrylates that are derived from low molecular epoxy resins and epoxy novolaks are known for their good thermal and mechanical properties as well as for their good resistance to aggressive chemicals. However, the tackiness and edge coverage of the resist films obtained with these systems on conductors owing to the fairly low relative molar mass are unsatisfactory. In practical application it is

Japanese patent Kokai Hei 04-294352 discloses the modification of epoxy-novolak resins by reaction with an unsaturated monocarboxylic acid and subsequently with an unsaturated anhydride of a polycarboxylic acid and the use thereof in photosensitive aqueous formulations. Furthermore, European patent application 0 292 219 discloses photosensitive systems that contain epoxy compounds of bisphenol A that are modified

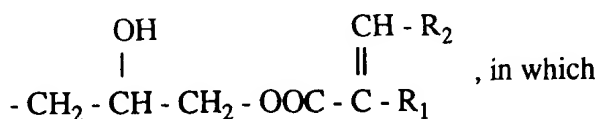
[illegible]
$$Q - \left[\begin{array}{c} \text{OM} \\ | \\ \text{O} - \text{A} - \text{O} - \text{CH}_2 - \text{CH} - \text{CH}_2 \text{O} - \text{T} - \text{OCH}_2 \text{CH} - \text{CH}_2 \\ | \\ \text{OM} \end{array} \right]_n - L \quad (\text{III})$$

Q is hydrogen or a group of formula



T is the radical of an aromatic bifunctional compound, and

$$\begin{array}{c} \text{O} \\ \diagup \quad \diagdown \\ -\text{CH}_2 - \text{CH} - \text{CH}_2 \end{array} \quad \text{or}$$

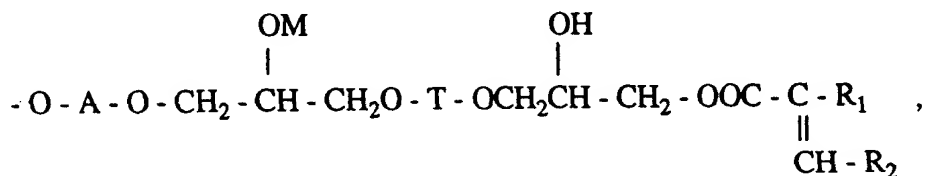
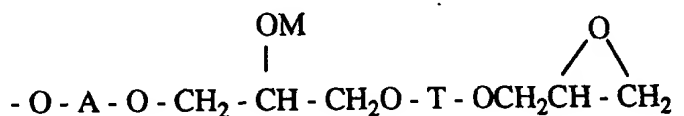


R_1 and R_2 are as defined above,

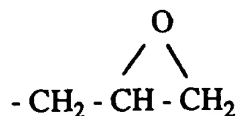
A is the radical of an aromatic bifunctional compound,

n is an integer from 0 to 300, and

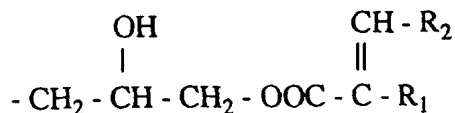
L is a group of formula



with the proviso that, in formula III, not all radicals M may simultaneously be hydrogen or a group of formula

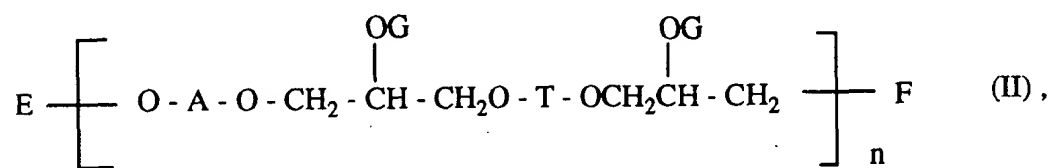


but at least 10 mol %, preferably 20-100 mol %, of the radicals M that are not present in the end groups Q and L denote a group of the above formula



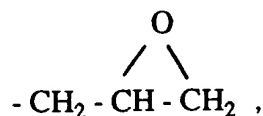
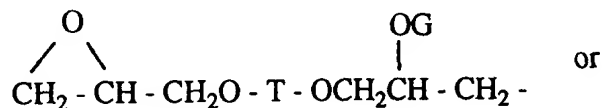
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The epoxy acrylates of formula III are obtained by reacting a postglycidylated epoxy resin of formula II

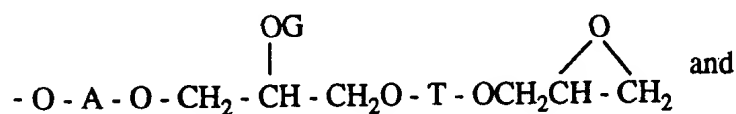


wherein

E is hydrogen or a group of formula

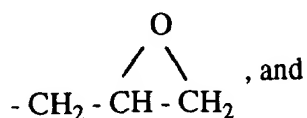


F represents the groups of formula - O - A - OG or



G is -H or the radical $\begin{array}{c} O \\ \diagup \quad \diagdown \\ - CH_2 - CH - CH_2 \end{array}$, and

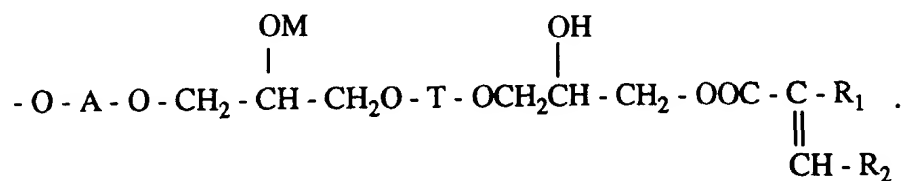
as in formula III, at least 10 mol % of the radicals G in formula II that are not present in the end groups E and F, represent the group of formula



A, T, and n have the given meanings,

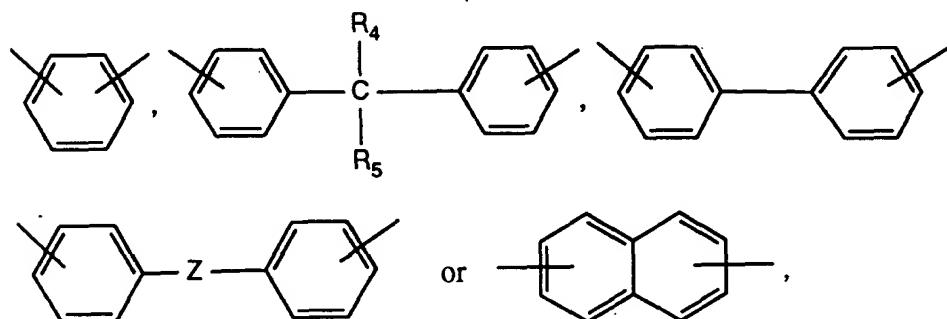
with an ethylenically unsaturated monocarboxylic acid in the presence of a catalyst and a polymerisation inhibitor, at elevated temperature.

If n in formula III is 0, then Q is -H and L is the group of formula



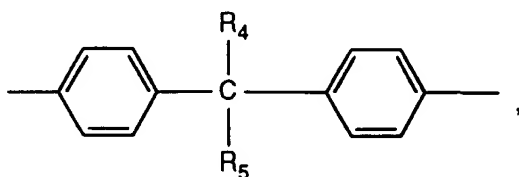
In preferred epoxy acrylates of formula III, n is an integer from 0 to 50, preferably from 0 to 30, and the symbols A and T have the preferred meanings of A and B given in Japanese patent Kokai Hei 1-195056.

Preferred bifunctional aromatic compounds A and T are linking groups of formulae

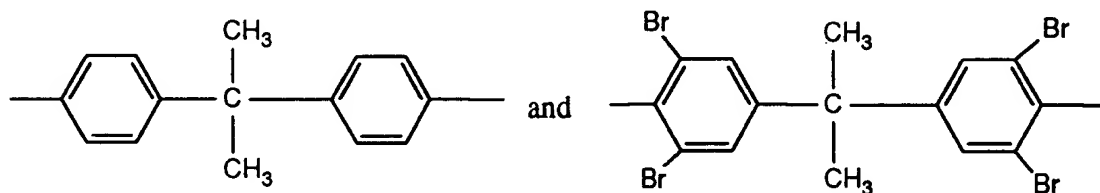


in which R_4 and R_5 are each independently of the other -H or C_1 - C_4 alkyl, Z is -S-, -O- or - SO_2 , and the aromatic radicals of the linking group A or T are unsubstituted or substituted by halogen or C_1 - C_4 alkyl. C_1 - C_4 Alkyl is preferably - CH_3 and halogen is preferably bromo.

Particularly preferred linking groups A and T each independently of the other have the formula

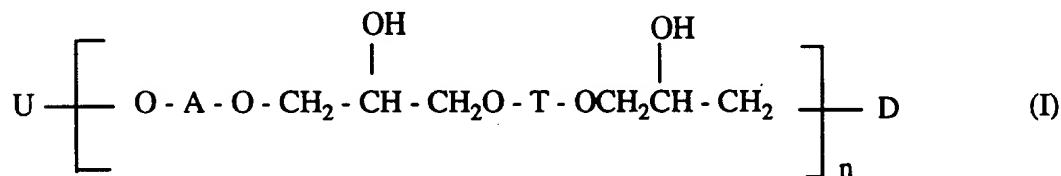


wherein R_4 and R_5 are as defined above and the phenyl radicals of the linking group are unsubstituted or substituted by bromo; and, preferably have the formulae



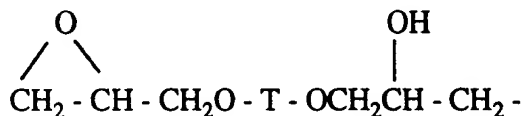
Throughout this specification, epoxy acrylates are taken to mean reaction products of epoxy compounds with (meth)acrylic acid.

Some of the postglycidylated epoxy resins of formula II are known and are prepared from the corresponding known advanced epoxy resins of formula I



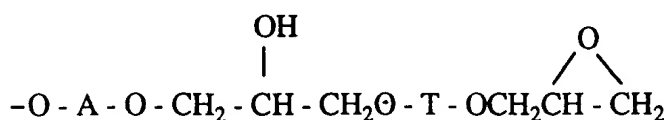
by a glycidylation reaction, wherein in formula I

U is hydrogen or the group of formula



and

D is the group of formula

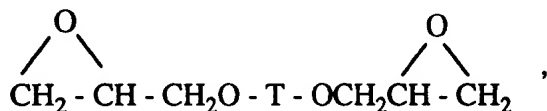


or the radical -O-A-OH, and the symbols A, T and n are as defined in connection with formula III.

The advanced epoxy resins of formula I on the other hand are obtained by known polyaddition of a bisphenol of formula



to a diepoxy of formula

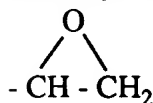


wherein A and T are the radical of a bifunctional aromatic compound.

The bisphenols HO-A-OH or HO-T-OH are preferably known bisphenols, more particularly bisphenol A and tetrabromobisphenol A, as well as those that are described in Japanese patent Kokai Hei 1-195056, preferably bisphenol A, bisphenol F, tetrabromobisphenol A and tetrabromobisphenol F.

To prepare the advanced epoxy resins, an excess of the above diepoxy is used, so that the advanced epoxy resins of formula I carry epoxy end groups. It is, however, also possible to use an excess of bisphenol HO-A-OH and thus to prepare molecules carrying phenolic end groups. The molecular weight is determined by the molar ratio of bisphenol HO-A-OH to the diepoxy. If desired, it is also possible to add minor amounts of higher functional phenols or epoxy compounds (e.g. trisphenols or triepoxyphenyl compounds) to the polyaddition. Specific amounts of starting materials (bisphenol HO-A-OH and/or diepoxy compound may also be present in the advanced epoxy resin.

The advanced epoxy resins of formula I contain secondary aliphatic hydroxyl groups resulting from the addition of the phenolic hydroxyl group to the oxirane ring



The glycidylation reaction to give the postglycidylated epoxy resins of formula II is

carried out by known methods by reacting the advanced epoxy resin (I) with e.g. an excess of epichlorohydrin in the presence of a base (e.g. NaOH) and of a catalyst at elevated temperature.

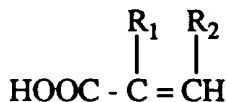
The amount of base used in the glycidylation will depend on the desired degree of glycidylation. It is preferred to use 0.1 to 1.2 equivalents of base per equivalent of secondary HO group in the advanced epoxy resin. Water is removed by azeotropic distillation using excess epichlorohydrin as entrainer.

Particularly suitable catalysts are quaternary ammonium salts or phosphonium salts, typically including tetramethylammonium chloride, benzyltrimethylammonium chloride, benzyltriethylammonium chloride, tetraethylammonium chloride and tetrabutylammonium bromide.

The reaction temperature is conveniently in the range from 40 to 80°C, preferably from 50 to 65°C.

The aliphatic OH groups are partially or completely glycidylated by the glycidylation reaction.

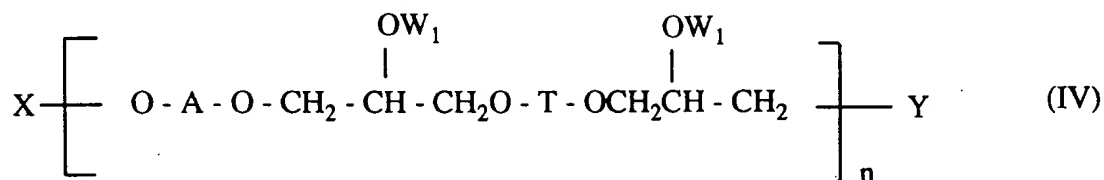
The further reaction of the postglycidylated epoxy resins of formula II to the ~~novel~~ epoxy acrylates of formula III is likewise carried out in known manner by reaction with an ethylenically unsaturated monocarboxylic acid of formula



Suitable acids typically include crotonic acid, cinnamic acid and, preferably, acrylic acid or methacrylic acid or a mixture thereof. R₁ and R₂ are as defined above.

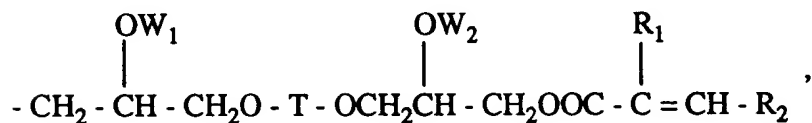
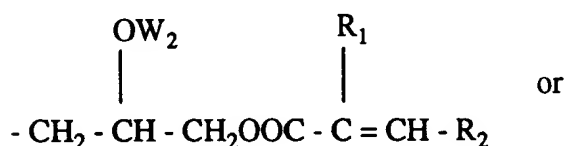
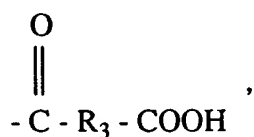
It is preferred to use a catalyst in the reaction. Particularly suitable catalysts are metal salts such as chromium compounds, amines such as triethylamine or benzyldimethylamine, also ammonium salts such as benzyltrimethylammonium chloride, or also triphenylphosphine and triphenylbismuth.

The completely acrylated epoxy acrylates of formula III can then be further reacted to carboxyl group-containing epoxy acrylates of formula IV



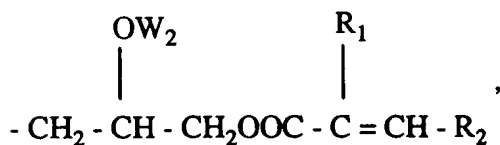
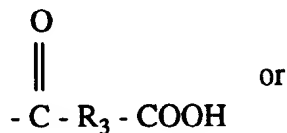
wherein

X is hydrogen or a group of formula



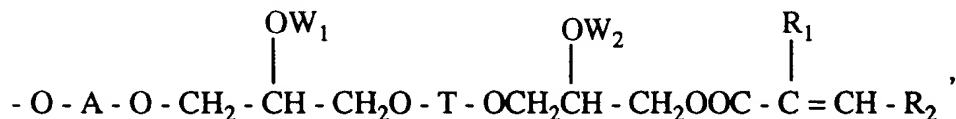
R₃ is the radical of a cyclic anhydride of a polycarboxylic acid after removal of the anhydride radical,

W₁ is hydrogen or a group of formula

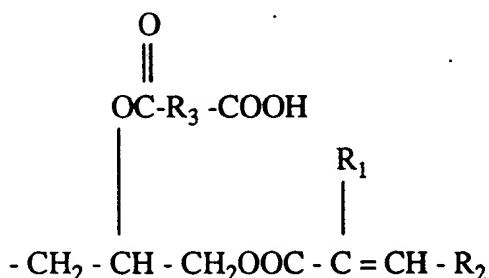


W₂ is -H or the group $\begin{array}{c} \text{O} \\ || \\ - \text{C} - \text{R}_3 - \text{COOH} \end{array}$, and

Y is the group of formula -O-A-O-W₁ or



wherein the symbols A, T, R₁, R₂, R₃ and n have the given meanings, with the proviso that, in formula IV, at least 10 mol % of radicals W₁ that are not in the end groups X and Y are a group of formula



R₁, R₂ and R₃ have the given meanings.

As the completely reacted epoxy acrylates of formula III contain almost no more epoxy groups, they can be reacted with cyclic anhydrides of polycarboxylic acids. In this case, the aliphatic hydroxyl groups react with the cyclic anhydride to effect ring opening and hemiester formation. In this reaction, for each reacted hydroxyl group a carboxylic acid bonded to the resin forms. The reaction comprises reacting the epoxy acrylate of formula III with the cyclic anhydride, in the absence or presence of a catalyst and of a polymerisation inhibitor, at elevated temperature. The HO groups of the compounds of formula II are completely or partially acylated, accompanied by ring opening of the anhydride. It is therefore advantageous that the epoxy acrylates contain no more epoxy groups, otherwise gelation occurs. The anhydride is used in equimolar amounts with respect to the hydroxyl groups or in a less than equivalent amount. The reaction is known per se.

Suitable anhydrides of polycarboxylic acids typically include succinic anhydride, maleic anhydride, glutaric anhydride, tetrahydrophthalic anhydride, hexahydrophthalic anhydride, 3-methyl- and 4-methylhexahydrophthalic anhydride, 3-ethyl- and 4-ethylhexahydrophthalic anhydride, 3-methyl-, 3-ethyl-, 4-methyl- and 4-ethyltetrahydrophthalic anhydride, itaconic anhydride, phthalic anhydride, and trimellitic anhydride. Preferred anhydrides are succinic, tetrahydrophthalic, hexahydrophthalic and phthalic anhydride.

Suitable catalysts typically include amines such as triethylamine, benzyldimethylamine, pyridine or dimethylaminopyridine, or triphenylphosphine or metal salts such as chromium or zirconium compounds.

If desired, a solvent may be added to the reaction medium, as the epoxy acrylates of formula III are in the form of solids. The solvent must, however, be inert to the cyclic anhydride, so that hydroxyl group-containing solvents are not suitable. The solvents cited in connection with the reaction with the ethylenically unsaturated monocarboxylic acids may suitably be used.

The reaction temperature is conveniently in the range from 60 to 140°C, and suitable polymerisation inhibitors are typically hydroquinone, hydroquinone monomethyl ether and 2,6-di-tert-butyl-p-cresol.

It is desirable to introduce dry air or a mixture of nitrogen/oxygen into the reaction medium. In a preferred embodiment of the invention, the epoxy acrylates of formula III are further reacted, without isolation, in the same reactor to the derivatives of formula IV modified with carboxyl groups.

Isolation and purification of the novel carboxyl group-containing epoxy acrylates of formula IV is usually not necessary. The reaction solution can be further used as obtained in the synthesis.

Owing to the unsaturated groups present in the molecule, the epoxy acrylates of formula III and the carboxyl group-containing epoxy acrylates of formula IV are thermally and photochemically crosslinkable. They can therefore be used and applied as acrylate components in photoresist formulations for the production of solder resists or primary resists by known methods, as for example disclosed in Swiss patent application 2005/93-4, filed on 2nd July 1993, entitled "Photopolymerisable compositions", and give resist layers having enhanced thermal, mechanical, electrical and chemical properties. The resist formulations prepared therefrom are used in particular in the field of printed wiring boards as solder resists or primary resists, and of printing plates. They are also suitable for the production of offset printing plates, flexographic printing plates, book printing plates and screen printing formulations. Suitable developers are aqueous as well as aqueous-organic or organic systems. Owing to the presence of carboxyl groups in the compounds of formula IV, these systems are particularly suitable for the preparation of aqueous-alkaline

developable photoresists.

Compared with low molecular epoxy acrylates in formulations that contain polymer binders, it is surprising that formulations containing epoxy acrylates of higher molecular weight without the addition of such polymer binders bring about an enhancement and not a loss of photosensitivity, and also that no increase in tackiness results. Furthermore, use of the formulations as solder resists results in improved edge coverage of the conductors. As no additional polymer binders are used in such formulations, further advantages accrue with respect to the thermal, mechanical and electrical properties and, in particular, to the resistance to chemicals of the resist compositions prepared therefrom. The novel epoxy acrylates of formula III and the carboxyl group-containing epoxy acrylates of formula IV have an increased glass transition temperature.

The invention is illustrated by the following non-limitative Examples.

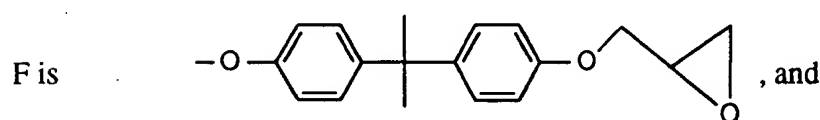
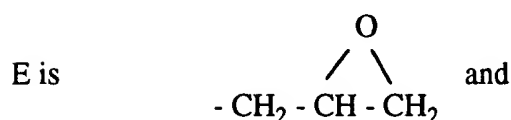
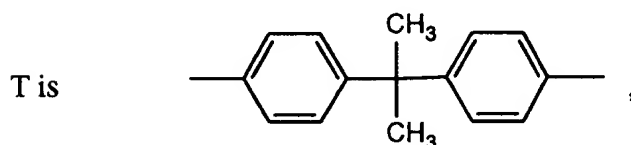
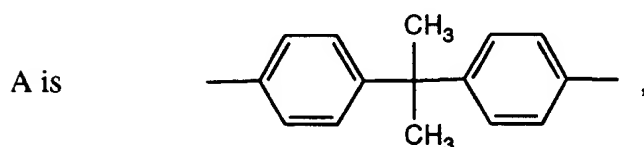
Preparative Examples

a) Postglycidylated epoxy resins (PGER)

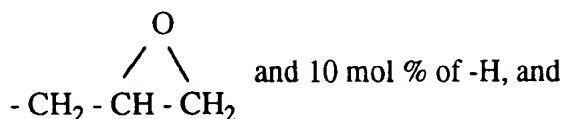
Example 1: The apparatus used is one suitable for tetramethylammonium chloride (TMAC) glycidylation reactions and which makes it possible to use excess epichlorohydrin as entrainer for the water separation under reduced pressure. The apparatus consists of a 5000 ml reactor equipped with stirrer, thermometer, an efficient multiple-coil condenser and 2 drip funnels with pressure equalisation. Connected to this apparatus are a water separator (for the upper aqueous phase and evacuation under vacuum) and a water jet pump with manometer. An oil bath is used for heating.

1000 g of epoxy resin Araldite GT 7004 (CIBA; (epoxy value 1.36 mol/kg of resin; HO number 2.71 mol) and 1700 ml (21.68 mol) of epichlorohydrin are dissolved in the reactor by heating. The homogeneous solution so obtained is heated to c. 80°C and, with efficient stirring, the water jet vacuum is cautiously applied such that the epichlorohydrin refluxes vigorously. The temperature in the reactor falls and is set to a constant temperature of c. 55°C and kept thereat for the entire reaction. This initially requires a pressure of 110-120 mbar, which towards the end of the reaction may then have to be somewhat reduced. The oil bath temperature is a constant 105-110°C. When the epichlorohydrin refluxes constantly at 55°C, 29.80 g (0.136 mol) of a 50 % aqueous solution of tetramethylammonium chloride are added rapidly from the first drip funnel. Then 195.20 g

(2.44 mol) of a 50 % aqueous solution of sodium hydroxide are added dropwise over 2 hours from the second drip funnel and simultaneously the water of reaction and the epichlorohydrin resulting from the 50 % solutions by azeotropic distillation are removed from the system. The internal temperature is kept at 55-58°C and sufficient reflux is maintained to ensure rapid removal of the water from the reaction mixture. After the dropwise addition, the reaction mixture is kept for 2 hours at 55°C while removing water from the system. The vacuum is discontinued and the suspension is neutralised by addition of crushed dry ice, followed by the addition of 10-20 ml of glacial acetic acid. After addition of 2000 ml of methoxypropyl acetate, the solvent and the epichlorohydrin are removed by distillation under reduced pressure. To the residue are added a further 2000 ml of methoxypropyl acetate and the resultant suspension is filtered over a filter aid (Hyflo), while scraping the surface of the filter bed from time to time. The clear, yellowish filtrate is concentrated on a rotary evaporator at a bath temperature of up to 120°C. The clear, yellow resin so obtained is diluted with methoxypropyl acetate to a solution of 50 % solids content, affording 2316 g of an almost colourless 50 % solution of the postglycidylated epoxy resin of formula II, wherein



G is c. 90 mol % of the group



n has a mean value of 2.

This solution can be used direct for the reaction with acrylic acid. The analytical data obtained for this solution are:

1. analysis (solids content): 49.5%;
2. epoxy value (titration): 1.46 mol/kg/(→ 2.81 mol/kg solid resin);
3. Cl content (solid resin): 0.40% Cl total;
0.23% Cl hydrolysable;
4. GPC (gel permeation chromatography polystyrene calibration) $M_w = 10678$; $M_n = 2138$.

If it is desired to obtain the solid resin, methoxypropyl acetate can be replaced as solvent with methyl isobutyl ketone for diluting the resin when working up and the residue, after concentration in the rotary evaporator, can be poured hot into flat steel pans and dried under a high vacuum at 150°C for several hours.

Example 2: Following the procedure described in Example 1, 555 g of Araldite GT 7004 (CIBA; (epoxy value 1.36 mol/kg of resin; HO number 1.50 mol), 705 ml of epichlorohydrin (9.00 mol), 16.40 g of a 50 % aqueous solution of tetramethylammonium chloride and 66.00 g of a 50 % aqueous solution of sodium hydroxide (0.83 mol) are reacted. After dropwise addition of the sodium hydroxide solution and further reaction for 45 min with separation of water, the bulk of the epichlorohydrin is removed by distillation under a vacuum of c. 85 mbar. Then 600 ml of methyl isobutyl ketone are added and the reaction mixture is neutralised with dry ice and acetic acid and diluted with a further 500 ml of methyl isobutyl ketone. The suspension is filtered over a filter aid (Hyflo), the solvent is stripped off on a rotary evaporator and the residue is poured hot into steel pans and dried under a high vacuum at 150°C for several hours. The following analytical data are

obtained for the yellowish solid resin (473 g):

1. epoxy value (titration): 2.39 mol/kg resin;
2. Cl content 0.33 % Cl total;
0.17% Cl hydrolysable;
3. OH number (titration): 1.22 mol/kg;
4. GPC (polystyrene calibration) $M_w = 10721$; $M_n = 2330$.

The solid resin conforms to formula II, wherein A and T as well as E, F and n have the meanings given in Example 1 and

G is c. 55 mol % of the group
$$\begin{array}{c} \text{O} \\ \diagup \quad \diagdown \\ -\text{CH}_2 - \text{CH} - \text{CH}_2 \end{array}$$
 and 45 mol % of -H.

Example 3: Following the procedure described in Example 1, 525 g of Araldite B 41 (CIBA; epoxy value 2.66 mol/kg of resin; HO number 1.01 mol), 712 ml of epichlorohydrin (9.09 mol), 18.48 g of a 50 % aqueous solution of tetramethylammonium chloride and 89.00 g of a 50 % aqueous solution of sodium hydroxide (1.11 mol) are reacted and the product is isolated as a solid resin.

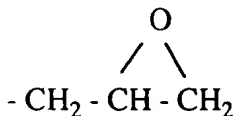
The analytical data obtained for the almost colourless solid resin (515 g) are:

1. epoxy value (titration): 3.48 mol/kg;
2. Cl content: 0.21% Cl total;
<0.05% Cl hydrolysable;
3. GPC (polystyrene calibration): $M_w = 2374$; $M_n = 935$.

The solid resin conforms to formula II, wherein T, E and F have the meanings given in Example 1, n is a mean value of c. 0.7 and

G is 100 mol % of

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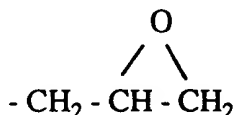


Example 4: Following the procedure described in Example 1, 435 g of Araldite GT 6071 (CIBA; epoxy value 2.18 mol/kg of resin; HO number 1.00 mol), 470 ml of epichlorohydrin (6.00 mol), 11.00 g of a 50 % aqueous solution of tetramethylammonium chloride and 88.00 g of a 50 % aqueous solution of sodium hydroxide (1.10 mol) are reacted and the product is isolated as a solid resin.

The analytical data obtained for the yellowish solid resin (390 g) are:

- | | |
|-----------------------------------|---|
| 1. epoxy value (titration): | 3.55 mol/kg; |
| 2. Cl content: | 0.82% Cl total;
0.60% Cl hydrolysable; |
| 3. GPC (polystyrene calibration): | $M_w = 6948$; $M_n = 1376$. |

The solid resin conforms to formula II, wherein A, T, E and F have the meanings given in Example 1, n is a mean value of c. 1.0 and G is 100 mol % of



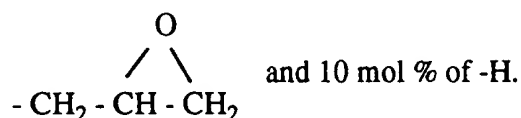
Example 5: Following the procedure described in Example 1, 1000 g of Araldite GT 6097 (CIBA; epoxy value 0.60 mol/kg of resin; HO number 3.175 mol), 1996 ml of epichlorohydrin (25.40 mol), 35.20 g of a 50 % aqueous solution of tetramethylammonium chloride and 228.60 g of a 50 % aqueous solution of sodium hydroxide (2.86 mol) are reacted and the product is isolated as a c. 50 % solution in methoxypropyl acetate.

The analytical data obtained for this solution (2283 g) are:

- | | |
|--------------------|---------|
| 1. solids content: | 48.30%; |
|--------------------|---------|

2. epoxy value (titration): 1.17 mol/kg;
3. Cl content: 0.63% Cl total;
0.30% Cl hydrolysable;
4. GPC (polystyrene calibration): $M_w = 16759$; $M_n = 3382$.

The solid resin in the solution conforms to formula II, wherein A, T, E and F have the meanings given in Example 1, n is a mean value of c. 5.3 and G is 90 mol % of

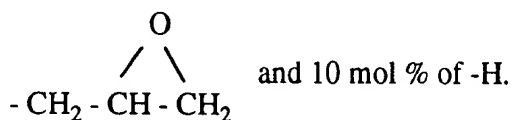


Example 6: Following the procedure described in Example 1, 500 g of Araldite GT 7097 (CIBA; epoxy value 0.58 mol/kg of resin; HO number 1.60 mol), 753 ml of epichlorohydrin (9.60 mol), 17.54 g of a 50 % aqueous solution of tetramethylammonium chloride and 115.20 g of a 50 % aqueous solution of sodium hydroxide (1.44 mol) are reacted and the product is isolated as a c. 45 % solution in methoxypropyl acetate.

The analytical data obtained for this solution (1200 g) are:

1. solids content: 44.50%;
2. epoxy value (titration): 1.19 mol/kg;
3. Cl content: 0.70% Cl total;
0.45% Cl hydrolysable;
4. GPC (polystyrene calibration): $M_w = 23091$; $M_n = 4270$.

The solid resin in the solution conforms to formula II, wherein A, T, E and F have the meanings given in Example 1, n is a mean value of c. 5.5 and G is 90 mol % of

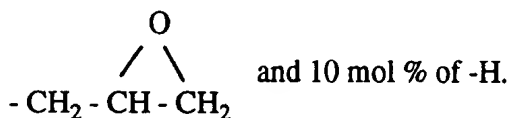


Example 7: Following the procedure described in Example 1, 400 g of Araldite GT 6099 (CIBA; epoxy value 0.41 mol/kg of resin; HO number 1.32 mol), 1038 ml of epichlorohydrin (11.88 mol), 14.08 g of a 50 % aqueous solution of tetramethylammonium chloride and 95.04 g of a 50 % aqueous solution of sodium hydroxide (1.19 mol) are reacted and the product is isolated as a c. 45 % solution in methoxypropyl acetate.

The analytical data obtained for this solution (915 g) are:

1. solids content: 45.60%;
2. epoxy value (titration): 1.16 mol/kg;
3. Cl content: 0.42% Cl total;
0.24% Cl hydrolysable;
4. GPC (polystyrene calibration): $M_w = 57017$; $M_n = 5604$.

The solid resin in the solution conforms to formula II, wherein A, T, E and F have the meanings given in Example 1, n is a mean value of c. 8.0 and G is 90 mol % of

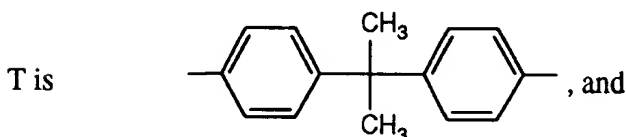
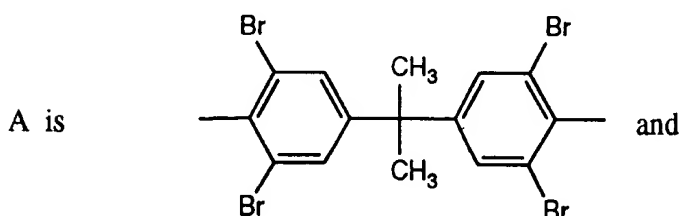


Example 8: Following the procedure described in Example 1, 1171 g of Araldite solid resin (CIBA; brominated epoxy resin; epoxy value 1.85 mol/kg of resin; HO number 2.26 mol), 1416 ml of epichlorohydrin (18.10 mol), 24.80 g of a 50 % aqueous solution of tetramethylammonium chloride and 162.60 g of a 50 % aqueous solution of sodium hydroxide (2.03 mol) are reacted and the product is isolated as a solid resin.

The analytical data obtained for this solid resin (1050 g) are:

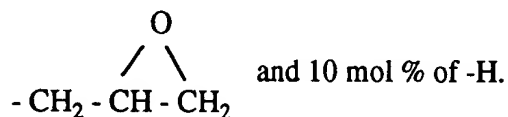
1. epoxy value (titration): 2.40 mol/kg;
2. Cl content: 0.45% Cl total;
0.22% Cl hydrolysable;
3. GPC (polystyrene calibration): $M_w = 3189$; $M_n = 1235$.

The solid resin conforms to formula II, wherein



E and F are as defined in Example 1 n is c. 1.0 and

G is 90 mol % of



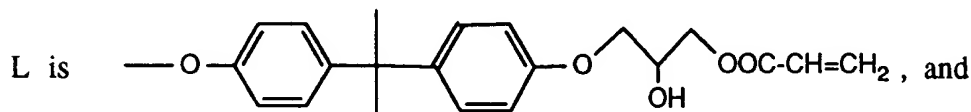
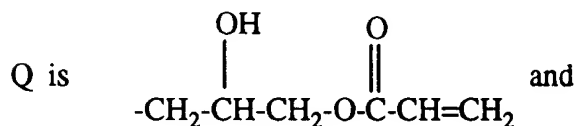
b) Partially or complete acrylated epoxy acrylates of PGER

Example 9: The apparatus used consists of a 5000 ml reactor equipped with stirrer, drip funnel, thermometer, reflux condenser and an inlet pipe for air. To inhibit the polymerization of the acrylates, a weak flow of air is introduced during the reaction below the surface. Heating is carried out with a thermostatically controlled oil bath. The reactor is charged with 2316.5 g of a 45 % solution (in methoxypropylacetate) of the reaction product of Example 1 (2.965 mol of epoxy groups), 3.80 g of hydroquinone monomethyl ether and 5.20 ml of triethylamine. With stirring, the reaction mixture is heated to 95°C. While introducing a weak stream of air, 213.70 g of acrylic acid (2.965 mol) are added

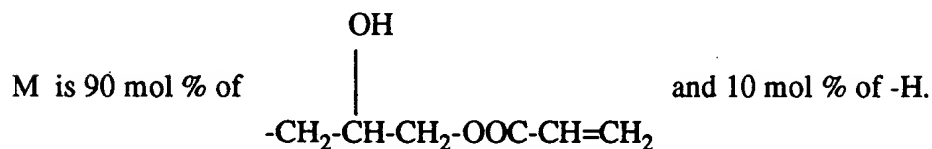
over 1 hour at 95°C and the reaction mixture is kept for 25 hours at this temperature. The reaction is monitored by measuring the epoxy value; upon completion of the reaction it is less than 0.08 mol/kg. The reaction product is cooled and further used without additional purification.

1. solids content: 54 %;
2. acid number (titration): 0.03 mol/kg;
3. viscosity (Brookfield) 25°C 1970 mPa.s;
4. GPC (polystyrene calibration): $M_w = 8889$; $M_n = 2256$.

The solid resin conforms to formula III, wherein A and T have the meanings given in Example 1,



n has a value of c. 2.0 and



Example 10: The apparatus used consists of a 2000 ml reactor equipped with stirrer, drip funnel, thermometer, reflux condenser and an inlet pipe for air. To inhibit the polymerisation of the acrylates, a weak flow of air is introduced during the reaction below the surface. Heating is effected with a thermostatically controlled oil bath. The reactor is charged with the following components and stirred at room temperature until a homogeneous solution has been obtained: 1018.60 g of a 50 % solution (in methoxypropyl acetate) of the reaction product of Example 1 (1.422 mol of epoxy groups titrated);

102.46 g of acrylic acid (1.422 mol); 1.53 g of hydroquinone monomethyl ether; 20.38 ml of a 10 % solution (in methoxypropyl acetate) of Nuosynchromium 5 (ex HARCROS, Durham Chemicals, Durham DH3 1QX, GB).

The reaction mixture is heated to 110°C and kept at this temperature with stirring and while introducing a weak flow of air for 7 hours. The reaction course is monitored by acid or epoxy titration. The acid titration number is 0.05 mol/kg after 7 hours. After cooling, the reaction product is drawn off and further used without purification. The analytical data obtained for the reaction product (1080 g) are:

1. solids content: 53.3 %;
2. epoxy value (titration): 0.02 mol/kg;
3. GPC (polystyrene calibration): $M_w = 12168$; $M_n = 2602$.

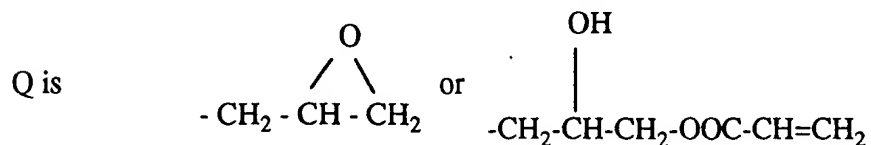
The solid resin conforms to the chemical structure of Example 9.

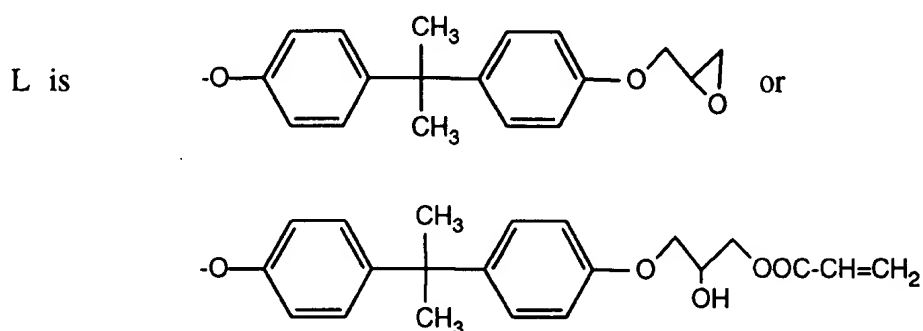
Example 11: The apparatus (500 ml reactor) as described in Example 10 is used.

Following the procedure described in Example 10, 200 g of a 50 % solution (in methoxypropyl acetate) of the reaction product of Example 1 (0.292 mol of epoxy groups), 10.52 g of acrylic acid (0.146 mol), 0.33 g of hydroquinone monomethyl ether and 20.00 ml of a 1 % solution (in methoxypropyl acetate) of Nuosynchromium 5 are reacted. The reaction is terminated after 3 hours at 100°C. The acid titration then gives <0.02 mol/kg. The analytical data obtained for the reaction product (210 g) are:

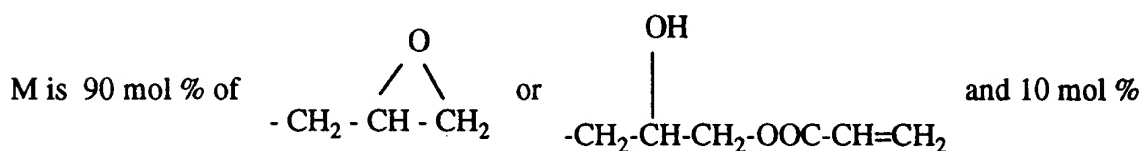
1. solids content: 46 %;
2. epoxy value (titration): 0.71 mol/kg;
3. GPC (polystyrene calibration): $M_w = 11036$; $M_n = 2352$.

The solid resin conforms to formula III, wherein A and T are as defined in Example 1,





and n is 2, and



of -H.

Example 12: The apparatus of Example 11 is used. Following the procedure described in Example 10, 200 g of a 50 % solution (in methoxypropyl acetate) of the reaction product of Example 5 (0.236 mol of epoxy groups), 17.00 g of acrylic acid (0.236 mol), 0.33 g of hydroquinone monomethyl ether and 0.54 g of triphenylphosphine are reacted. The reaction is terminated after 14 hours at 100°C. The acid titration then gives 0.09 mol acid/kg. The analytical data obtained for the reaction product (210 g) are:

1. solids content: 54 %;
2. epoxy value (titration): 0.10 mol/kg;
3. GPC (polystyrene calibration): $M_w = 19493$; $M_n = 3075$.

The solid resin conforms to formula III, wherein A and T are as defined in Example 1, and Q, L and M are as defined in Example 9, and n is c. 5.3.

Example 13: The apparatus described in Example 11 is used. Following the procedure described in Example 10, 100 g of a 48.3 % solution (in methoxypropyl acetate) of the reaction product of Example 5 (0.117 mol of epoxy groups), 8.85 g of acrylic acid (0.123 mol), 0.15 g of hydroquinone monomethyl ether and 1.15 ml of a 10 % solution (in

methoxypropyl acetate) of Nuosynchromium 5 are reacted. The reaction is complete after 7 hours at 110°C. The acid titration then gives 0.06 mol/kg. The analytical data obtained for this reaction product (210 g) are:

epoxy value (titration):	0.020 mol/kg.
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The product conforms to the formula of Example 12.

Example 14: The apparatus (1000 ml) described in Example 10 is used. Following the procedure described in Example 10, 720 g of a 50 % solution (in methoxypropyl acetate) of the reaction product of Example 5 (0.828 mol of epoxy groups), 29.83 g of acrylic acid (0.414 mol), 1.11 g of hydroquinone monomethyl ether and 37.00 ml of a 2 % solution (in methoxypropyl acetate) of Nuosynchromium 5 are reacted. The reaction is complete after 5 hours at 100°C. The acid titration then gives <0.02 mol acid/kg. The analytical data obtained for this reaction product are:

- | | |
|-----------------------------------|--------------------------------|
| 1. solids content: | 50.3 %; |
| 2. epoxy value (titration): | 0.65 mol/kg; |
| 3. GPC (polystyrene calibration): | $M_w = 38840$; $M_n = 4226$. |

The solid resin conforms to formula III, wherein A and T are as defined in Example 1, and Q, L and M are as defined in Example 11, and n is c. 5.3.

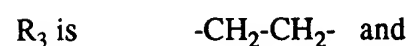
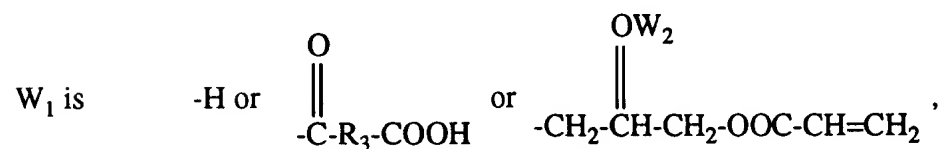
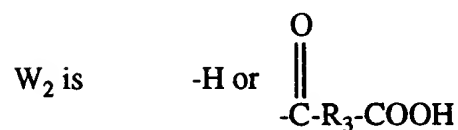
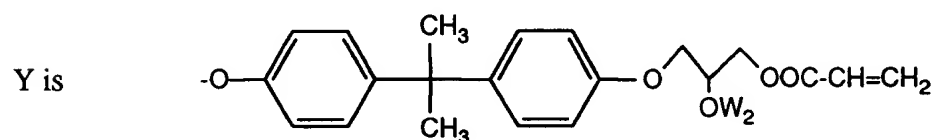
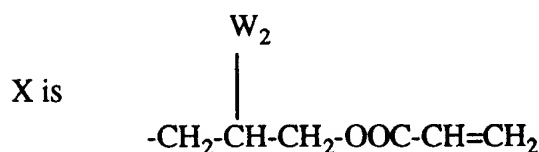
c) Carboxyl group-containing epoxy acrylates of completely acrylated PGER

Example 15: The apparatus used consists of a 5000 ml reactor equipped with stirrer, thermometer, reflux condenser and an inlet pipe for air. To inhibit the polymerisation of the acrylates, a weak flow of air is introduced during the reaction below the surface. Heating is carried out with a thermostatically controlled oil bath. The reactor is charged with 2171 g of the reaction product of Example 9 (2.61 mol of OH groups) and, with stirring, 196.08 g of succinic anhydride (1.96 mol) are added. Then 7.4 ml of benzyldimethylamine are added and the reaction mixture is heated to 110°C while introducing a flow of dry air and kept at this temperature for 4 hours. Then a further 7.4 ml of benzyldimethylamine are added and stirring is continued for 4 hours at this temperature.

The homogeneous yellowish reaction product is cooled and further used without additional purification. The following analytical data are obtained for this reaction product:

1. solids content: 55 %;
2. acid number (titration): 0.88 mol/kg;
3. GPC (polystyrene calibration): $M_w = 8525$; $M_n = 2237$.

The solid resin conforms to formula IV, wherein A and T have the meanings given in Example 1, and



n is 2.0.

Example 16: The apparatus (500 ml) as described in Example 15 is used. The reactor is charged with the reaction product of Example 13 (0.117 ml OH groups) and 5.854 g of

succinic anhydride (0.0585 mol), and the mixture is heated to 110°C and kept at this temperature for 7 hours while introducing a weak flow of air. The homogeneous reaction product is further used without additional purification. The following analytical data are obtained for this reaction product:

1. solids content: 61.7 %;
2. acid number (titration): 0.66 mol/kg.

The solid resin conforms to the structure of Example 15 and n is c. 5.3.

Application Examples

General procedure: The coating substrates used are cleansed copper-clad electronic laminates or processed printed circuit boards with conductive pattern. The resist formulations are prepared by mixing and dissolving the components listed in the Examples, followed by optional filtration. All operations are carried out under protective yellow light.

For test purposes, the formulations can be coated on to the printed circuit board with a wire applicator. For larger series, curtain coating methods or roller coating as well as screen printing are used.

Drying is carried out in a circulating air oven. Exposure is made using a commercial apparatus with 5000 W mercury high-pressure radiation source doped with metal halide. Development is carried out in commercial continuous development apparatus. Assessment of photosensitivity and resolution is made by exposure through a Stouffer step wedge and resolution wedge and the result is evaluated from the developed resist image.

Application Example

Formulation 1.1 comprises a novel acrylate modified with carboxyl groups:

53.00 g	of the reaction product of Example 15 (as 55 % solution in methoxypropyl acetate);
19.00 g	of CN 965 (acrylate sold by Craynor);
3.00 g	Irgacure 907 (photoinitiator; CIBA AG, Basel);
1.50 g	Quantacure ITX (isopropylthioxanthone; sensitiser);

0.15 g Orasol blue GN (dye sold by CIBA AG, Basel);
28.95 g methoxypropyl acetate.

Formulation 1.2: Comparison Example with a highly polymerised binder and without novel acrylate:

97.20 g of Scripset 550E solution (30 % solution in methoxypropyl acetate; polymer binder sold by Monsanto; styrene/maleic acid copolymer);
19.00 g of CN 965 ;
3.00 g of Irgacure 907;
1.50 g of Quantacure ITX;
0.15 g of Orasol blue GN.

Result:

	Formulation 1.1	Formulation 1.2
solids content	50 %	43.7 %
viscosity 25°C (Epprecht)	600 mPa.s	7600 mPa.s
dry layer thickness	12µm	12µm
drying conditions	5 min. 80°C	5 min. 80°C
exposure	150 mJ/cm ²	150 mJ/cm ²
development	1 % Na-carbonate 35°C	1 % Na-carbonate 35°C
last visible wedge step	6	no image: developed away

Formulation 1.1 containing the novel acrylate has a significantly lower viscosity than formulation 1.2, despite the substantially higher solids content. In addition, the photosensitivity (Stouffer wedge step 6) is much higher compared with formulation 1.2 (a too low sensitivity to produce an image).